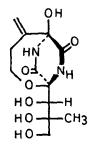
TOTAL SYNTHESIS OF BICYCLOMYCIN¹

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Summary: Total synthesis of (\pm) -bicyclomycin was achieved in 19 steps starting from diketopyperazine.

Bicyclomycin is an antibiotic found in Japan. It was produced by <u>Streptomyces sapporonensis</u> and <u>S</u>. <u>aizunensis</u>² and has a unique spectrum of antibacterial activities. Structure of bicyclomycin (1) has been elucidated by chemical and X-ray analyses.² It has a novel bicyclo[4.2.2] system containing oxidized diketopiperazine ring and a side chain at its bridge-head position. Several reports³ directing to its synthesis have been appeared, but no total synthesis of bicyclomycin has been reported. We have reported synthesis of (±)-N,N',O-trimethylbicyclomycin in the previous papers.^{4,5} This paper deals with a total synthesis of (±)-bicyclomycin.



Bicyclomycin (1)

Diketopyperazine (2) was protected by benzyl group⁶ (3, 65% yield) and then brominated with bromine in chlorobenzene at 130 °C (4, 92% yield). On heating with benzyl alcohol the dibromide 4 afforded the dibenzyl ether 5 in 85% yield as a 3:1 mixture of cis (mp 60-61 °C) and trans (mp 148 °C) form, which were separated by fractional crystallization. Each isomer or the mixture was used for the next step.

Methyl γ -hydroxycrotonate (\oint_{γ}), the component for the side chain, was prepared from methyl crotonate by oxidation with selenium dioxide followed by reduction with NaBH₄. Its hydro-xy group was protected with <u>t</u>-butyldiphenylsilyl group (TBDPS) by treatment with TBDPS chloride and imidazole to

give methyl γ -TBDPS-oxycrotonate (7), mp 77-78 ∞^7 (76% yield).

The monoanion prepared from \S and <u>n</u>-BuLi in THF at -110 °C was treated with χ at -78 °C. This conjugate addition proceeded stereospecifically and gave only one condensed product \S . Stereochemistry of \S on the diketopiperazine ring was as shown.⁹ It was reduced with LiAlH₄ in THF at -78 °C to yield the crude aldehyde \S [PMR⁹ 8.22 (1H, br.t, CHO)], which was further reduced with NaBH₄ in methanol followed by silica gel column chromatography to give the alcohol $\sharp\emptyset$, mp 121-122 °C⁷ (39% yield from \S). The primary hydroxy group of $\sharp\emptyset$ was protected with <u>t</u>-butyldimethylsilyl group (TBDMS) by treatment with TBDMS chloride and imidazole to $\sharp\natural$ almost quantitatively [MS m/z 889; PMR⁹ -0.08 (6H, s), 0.86 (9H, s), 1.02 (9H, s)].

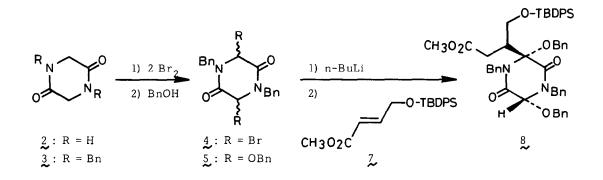
In the synthesis of N,N',O-trimethylbicyclomycin the secondary methoxy group corresponding to the secondary benzyl group in 11 was displaced smoothly by acetoxy group by treatment with $Ac_{2}O/CF_{3}CO_{2}H$,⁴ but the secondary benzyl group in 11 could not be displaced by acetoxy group under the similar condition. Therefore, 11 was hydrogenated in ethanol in the presence of 20% Pd-C and pyridine (5 v/v%) at 40 °C. Only one benzyl group (secondary rather than tertiary) was selectively removed and the mono-debenzylated product 12 was crystallized from hexane (88% yield), mp 136 °C⁷ [PMR⁹ 5.16 (1H, d — s by addition of D₂O), 5.74 (1H, d — disappeared by addition of D₂O)]. Acetylation of 12 afforded the acetate 13 (quantitative yield) [MS m/z 841; PMR⁹ 1.92 (3H, s)], whose TBDMS group was selectively removed to give the mono-ol 14 (80% yield) [MS m/z 769; no PMR signal around -0.1 ppm]. It was heated in dichloroethane at 80 °C in the presence of pyridinium tosylate, affording the bicyclo compound 15¹⁰ (84% yield), mp 122-123 °C.⁷

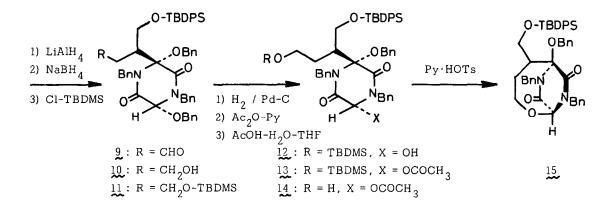
Aldol condensation of the bicyclo compound 15 and the aldehyde 16 would give four stereoisomers. We have studied the stereoselectivity of this type of condensation⁵ and found that the major isomer was formed in more than 50% of the four stereoisomers and that it had the same relative stereochemistry of that of bicyclomycin ($\frac{1}{2}$).

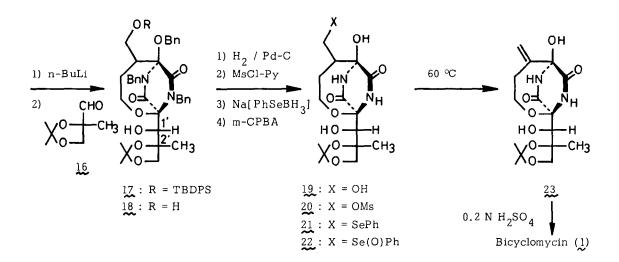
The bicyclo compound 15 in THF was converted to its monoanion with <u>n</u>-BuLi at -110 °C and condensed with (±)-2-methylglyceraldehyde acetonide(16)¹¹ at -78 °C to give a mixture of four stereoisomers in a 3:1:1:0 ratio from which the major product 17, mp 186 °C, ⁷ was isolated in 41% yield by means of preparative tlc. This was treated with 1M <u>n</u>-Bu₄NF in THF to give the primary alcohol 18, mp 177-178 °C, ⁷ almost quantitatively. All of the benzyl protecting groups of 18 were removed by catalytic hydrogenation in ethanol under H₂ atmosphere in the presence of 20% Pd-C at 80 °C for 12 h to yield the completely debenzylated product 19 (57% yield), mp 216-218 °C (dec). ⁷ Mesylation of 19 afforded the monomesylate 20 (83% yield) [MS m/z 423; PMR ⁹ 3.07 (3H, s)], which in abs. ethanol under N₂ atmosphere was treated at room temperature with 0.5M Na[PhSeBH₃] solution prepared from NaBH₄ and PhSeSePh in ethanol. The selenide 21 was isolated in 40% yield [MS m/z 500, 498; PMR ⁹ 7.1-7.6 (5H, m)]. Oxidation of 21 with <u>m</u>-chloroperbenzoic acid in dichloromethane gave the selenoxide 22, which , after chromatography on a silica gel column, was dissolved in dichloroethane and the solution was heated at 60 °C for 20 min to give bicyclomycin acetonide (23) (69% yield), mp 164-170 °C (dec). ⁷

The acetonide 23 was hydrolyzed by careful treatment with 2 equivalents of 0.2N sulfuric acid initially at 0 $^{\circ}$ and then at 25 $^{\circ}$ for 8 h.⁵ Purification using a silica gel column and an ODS RP-18 column, and crystallization from methanol-acetone afforded (±)-bicyclomycin (1) (66% yield), mp 166-170 $^{\circ}$ C, whose PMR and CMR spectra¹⁴ as well as Rf values on the were completely identical with those of natural bicyclomycin.

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- 3. (a) H. Maag, J. F. Blount, D. L. Coffen, T. V. Steppe, and F. Wong, J Am. Chem. Soc., <u>100</u>, 6786 (1978); (b) L. V. Dunkerton and R. W. Ahmed, Tetrahedron Lett., <u>21</u>, 1803 (1980); (c) C. Shin, Y. Sato, and J. Yoshimura, Tetrahedron Lett., <u>22</u>, 2401 (1981); (d) T. Fukuyama, B. D. Robins, and R. A. Sachleben, Tetrahedron Lett., <u>22</u>, 4155 (1981); (e) J. H. Hoare and P. Yates, J. Chem. Soc., Chem. Commun., 1126 (1981); P. Yates and J. H. Hoare, Can. J. Chem, <u>61</u>, 519 (1983); (f) R. M. Williams, O. P. Anderson, R. Armstrong, J. Josey, H. Meyers and C. Eriksson, J. Am. Chem. Soc., <u>104</u>, 6092 (1982); (g) R. M. Williams, J. S. Dung, J. Jesey, R. W. Armstrong and H. Meyers, J. Am. Chem. Soc. <u>105</u>, 3214 (1983).
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- 5. S. Nakatsuka, K. Yoshida and T. Goto, Tetrahedron Lett., 22, 4973 (1981).
- <u>p</u>-Methoxybenzyl group may be a better protecting group than benzyl group for easy removal, but in this case it cannot be used, for it is susceptible toward bromination.
- 7. Satisfactory elemental analyses were obtained.
- 8. S. Nakatsuka, K. Sasaki, K. Yamaguchi and T. Goto, Chemistry Lett., 695 (1981).
- 9. PMR spectra were taken in CDCl_3 at 100 MHz and the chemical shifts are in $\delta(\text{ppm})$.
- 10. PMR⁹ of 15: 0.98 (9H, s), 1.62 (2H,m), 1.7-2.4 (1H,m), 3.20-3.64 (2H,m), 3.90 (1H,d,J=10 Hz), 4.06 (1H,d,J=14 Hz), 4.3 (1H,m), 4.46 (1H,d,J=15 Hz), 4.60 (1H,d,J=10 Hz), 4.74 (1H, d,J=15 Hz), 5.12 (1H,d,J=14 Hz), 5.16 (1H,s), 6.9-7.7 (25H,m).
- 11. Ref. 3(a).
- 12. Compound λ_{λ} could not be debenzylated completely by catalytic hydrogenation owing to the steric bulkiness of t-butyldiphenylsilyl group.
- 13. PMR of 23 (CD₃OD) ppm : 1.38 (3H, s), 1.43 (3H, s), 1.45 (3H, s), 2.64 (2H, m), 3.74 lH, d, J = 8 Hz), 3.89 (2H, m), 4.15 (1H, s), 4.45 (1H, d, J = 8 Hz), 5.14 (1H, d, J = 1.5 Hz), 5.58 (1H, d, J = 1.5 Hz); CMR (CD₃OD) ppm: 25.0, 26.8, 28.2, 36.5, 66.5, 73.1, 82.8, 86.4, 88.9, 111.5, 116.7, 149.1, 168.3, 172.0.
- 14. PMR of 1 (CD₃OD) ppm: 1.34 (3H, s), 2.60 (2H, m), 3.52 (1H, d, J = 11 Hz), 3.65 (1H, d, J = 11 Hz), 3.86 (2H, m), 4.08 (1H, s), 5.13 (1H, d, J = 1 Hz), 5.56 (1H, d, J = 1 Hz), CMR (CD₃OD) ppm: 24.2, 36.7, 65.5, 68.4, 72.0, 78.2, 82.9, 89.4, 119.8, 149.4, 168.6, 172.4.

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